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QSARs OF SOME ANTIBACTERIAL ACTIVE BENZOXAZOLES AGAINST B. SUBTILIS

BAZI ANTİBAKTERİYAL ETKİLİ BENZOKSAZOLLERİN *B. SUBTİLİS'* E KARŞI KANTİTATİF YAPI-ETKİ İLİŞKİLERİ

Özlem TEMİZ-ARPACI, İlkay YILDIZ-ÖREN, Esin AKI-ŞENER^{*}, İsmail YALÇIN and Betül TEKİNER

Ankara University, Faculty of Pharmacy, Department of Pharmaceutical Chemistry, 06100 Tandogan ANKARA-TURKEY

ABSTRACT

The QSAR analysis of a set of previously synthesized 5-substitutedbenzamido- and 5substitutedphenylacetamido-2-(p-substituted-phenyl)benzoxazole derivatives, which were tested in vitro for their growth inhibitory activity against Bacillus subtilis, was performed by using the stepwise multiple regression analysis. The resulting QSAR revealed that the substitution at position R_2 is more significant than R and R_1 to improve the antibacterial activity. Hydrophobic and steric effects of substituents at R_2 have an important role for increasing the antibacterial activity compared to other parameters.

Key words: QSAR, Antibacterial activity, Benzoxazoles

ÖZET

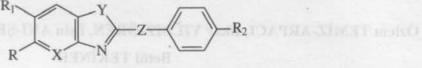
Bu kantitatif yapı-etki ilişkileri analizinde önceden sentezleri gerçekleştirilmiş ve B. subtilis'e karşı in vitro gelişimlerini inhibe etme aktiviteleri test edilmiş olan 5-sübstitüebenzamido- ve 5sübstitüefenilasetamido-2-(p-sübstitüefenil)benzoksazol türevlerine basamaklı çoklu regresyon analizi uygulandı. Kantitatif yapı-etki ilişkileri analiz sonuçları, R_2 konumunun R ve \hat{R}_1 'den antibakteriyal aktivite için daha önemli olduğunu ortaya koymuştur. R_2 'nin hidrofobik ve sterik etkileri antibakteriyal aktivitenin artması için diğer parametrelerden daha önemlidir.

Anahtar Kelimeler: kantitatif yapı-etki ilişkileri, antibakteriyal etki, benzoksazoller

Corresponding author Tel: +90(312)223 69 40 Fax: +90(312)223 69 40 e-mail:sener@pharmacy.ankara.edu.tr 74 Özlem TEMİZ-ARPACI, İlkay YIILDIZ-ÖREN, Esin AKI-ŞENER, İsmail YALÇIN, Betül TEKİNER

INTRODUCTION

In the last few years, we reported the synthesis and the antimicrobial activity of various 2,5-disubstituted benzoxazoles, benzimidazoles, benzothiazoles and oxazolo[4,5-b]pyridines (Figure 1), against some Gram-positive, Gram-negative bacteria and the yeast *Candida albicans*, providing a wide variety of *in vitro* antimicrobial effects especially indicating significant activity against the enterobacter *Pseudomonas aeruginosa* and the yeast *C. albicans* (1,2).



X; =CH-, =N-Y; -O-, -S-, -NH-Z; -CH₂-, -OCH₂-, -SCH₂-, -C₂H₄-R; -H, -Cl, -CH₃, -NO₂, -NH₂ R₁; -H, -CH₃, -NO₂ R₂; -H, -Cl, -F, -Br, -CH₃, -NO₂, -NH₂, -C₂H₅, -C(CH₃)₃, -OCH₃, -NHCH₃, -NHCOCH₃, -N(CH₃)₂

Figure 1. Previously synthesized 2,5,6-trisubstituted-benzoxazoles, benzimidazoles, benzothiazoles and oxazolo[4,5-b]pyridines.

The determination of the structure-activity relationships of *in vitro* antibacterial and antimycotic activities of the previously synthesized compounds revelead that these related fused heterocyclic systems generally behaved bioisosterically for the screened microorganisms. However, oxazolo[4,5-b]pyridine derivatives showed the best inhibitory potency for the *Klebsiella pneumoniae* and *C. albicans* (3-6).

In order to describe the nature of the interactions at the molecular level, developed QSAR analysis by using the quantum-chemical calculations revelead that the electrophilic superdelocalizability of the nitrogen atom in the oxazolo moiety of the benzoxazole ring and the lowest unoccupied molecular orbital energy levels of the compounds were found in relation with the activity and the fused heterocyclic system was found as the most important part in the molecule for the interactions (5,7).

In the present paper, a set of previously synthesized 2-(p-substituted-phenyl)- 5substituted-benzamido- and 5-substituted-phenyl-acetamidobenzoxazole derivatives **1-23** were tested for *in vitro* growth inhibitory activity against *B. subtilis* and the QSARs were analyzed by multiple regression analysis (MRA) in order to predict the lead optimization in this set of compounds.

Methodology

The Hansch analysis method has been most widely and effectively used for lead optimization in theoretical drug design. (9,10).

This method can be formulated as given in Eq. 1:

 $\log 1/C = \Sigma a_i I_i + \Sigma b_i X_i + c$

Eq. 1

where, I_i is the structural indicator parameters and X_i is the physicochemical variables.

In this study, the model is based on the *in vitro* activity of certain 2,5-disubstitutedbenzoxazole derivatives **1-23** (Table 1) against *B. subtilis*, where C is the molar concentration of the MIC values of the compounds.

The candidate set of descriptors used in this analysis were π as hydrophobic, σ , *F* and *R* as electronic and MW, MR, Es, *L*, *B*₁, and *B*₄ as steric parameters for the substituents R₁ and R₂ (11). Besides these physicochemical variables, structural indicator parameters were also taken into consideration for the substituents Y and R.

The QSAR analysis was performed by using the multiple regression technique and a nonlinear (parabolic) correlation was obtained between antibacterial activity and the lipophilic character of the substituents at position $R_2(12)$.

On the other side, the predictive power of the performed QSAR model was also determined by using the Cross-Validation Method (13-15).

Regression analysis and calculations were run on PC using the BILIN statistical program which was prepared by Hugo Kubinyi (16). In equations, the figures in parenthesis are the standard errors of the regression coefficients. For a given equation, n is the number of compounds, R^2 denotes the square of the multiple correlation coefficients, F is the significance test, s represents the residual standard deviation, Q^2 is the squared cross-validation regression coefficient and s-PRESS showes the standard deviation of cross-validation predictions.

In vitro microbiological activity

The antibacterial activities against the strain *B. subtilis* ATCC 6033 were determined as the minimum inhibitory concentration (MIC) values in vitro by a two-fold serial dilution technique (17,18). The test was performed using the compounds which were dissolved in absolute ethanol (0.4 mg/ml) and further control dilutions in the test medium were furnished at the required quantities of 400, 200, 100, 50, 25, 12.5, 6.25, 3.12, 1.56, 0.78 μ g/ml concentrations. In order to ensure that the solvent per se had no effect on bacterial growth, a control test was also performed containing inoculated broth supplemented with only ethanol at the same dilutions used in our experiments and found inactive in culture medium.

For the antibacterial assay, the cultures were obtained in Mueller-Hinton broth (Difco) for all the bacteria after 24 h of incubation at 37 \pm 1°C. Testing was carried out in Mueller-Hinton broth at pH 7.4 and the two-fold serial dilution technique was applied. The final inoculum size was 10⁵ CFU/ml. A set of tubes containing only inoculated broth was kept as controls. After incubation for 24 h at 37 \pm 1°C, the last tube with no growth of microorganism was recorded to represent MIC expressed in µg/ml. The potency has been defined as log 1/C in the QSAR analysis where *C* is the molar MIC value of the compounds. MIC and the observed log 1/C values of the tested compounds are listed in Table 1.

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Table 1: Compounds and parameters used in Eq 2 and 3.

Physicochemical parameters Parabolic Model Comp. Y MIC R R_1 R_2 πR_2 $B_4 R_2$ Obs. Cal. Residuals No: µg/ml Log1/C Log1/C H 0 3.895 -0.097 H Η 50 3.798 1 3.817 3.869 -0.052 H CH₃ 0.56 2.04 50 H H 3.690 0.125 H 1.02 2.97 50 3.815 3 C2H5 3.254 3.304 -0.050 NO2 -0.28 2.44 H H 200 H H C(CH₃)₃ 1.98 2.97 100 3.568 3.561 0.007 5 1.4 -0.060 C2H5 H 0 50 3.835 3.895 H 6 C2H5 C2H5 1.02 2.97 50 3.869 3.707 0.162 Η -3.895 H 50 3.843 -0.052 H CL 0 H OCH₃ OCH₃ -0.02 2.87 200 3.272 3.363 -0.091 9 H CH₃ 0.56 2.04 25 4.136 3.869 0.267 10 CH₃ -CH₂ 50 3.817 3.895 -0.078 11 Н H H 0 12 CH₂ Н Br 1.95 50 3.911 3.969 -0.058 Н 0.86 13 CH₂ H 0.71 1.8 50 3.860 3.979 -0.119 H CL 14 NO2 -0.28 200 3.271 3.304 -0.033 CH₂ Н H 2.44 OC₃H₇ 1.05 4.30 200 3.286 3.340 -0.054 15 CH₂ H H 16 50 3.853 3.895 -0.042 CH₂ C2H5 H Η 0 3.940 4.135 -0.195 CH₂ C2H5 Br 0.86 1.35 50 17 H 1.8 3.893 3.979 -0.086 18 CH₂ C2H5 Cl 0.71 50 Η 19 3.895 3.969 -0.074 H H Br 0.86 1.95 50 20 50 3.843 3.979 -0.136 Η H Cl 0.71 1.8 -3.895 0.225 21 Br 25 4.120 F H 0 22 1.8 25 4.180 3.979 OCH₂ $N(CH_3)_2$ Н Cl 0.71 0.201 25 3.995 0.205 23 SCH₂ N(CH₃)₂ H BH 0 4.200

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RESULTS AND DISCUSSION

As a result of QSAR analysis, Eq. 3 was obtained as the best equation for the lead optimization predictions in this set of tested compounds (Table 2). According to the applied stepwise regression technique and the validation test results the performed parabolic correlation equation model given as below.

 $\log 1/C = -0.254 (\pm 0.16) [\pi]^2 R_2 + 0.611 (\pm 0.25) \pi R_2$

- 0.278 (±0.090) $B_4 R_2$ + 4.173 (±0.17) Eq. 3 π -optimum = 1.20

n = 23; R²= 0.872; s = 0.149; F = 20.114; p<0.001 Q² = 0.670; s-PRESS = 0.175

The correlation coefficients which are given in Table 3 reveal that there is no collinearity between the independent variables used in eq.3.

Compounds and the parameters used in this QSAR analysis together with the observed, calculated and residual values are given in Table 1.

QSAR analysis reveals that the substitution at position R_2 is significant rather than the position R, R_1 and Y for the tested antibacterial activity. Substituting position R_2 with a group which has a hydrophobic character possessing a π value of 1.20 increases the activity. Additionally, it has also found that substituent having a maximum width at this position enhances the activity against *B. subtilis*.

q. No.	Equation	n	R ²	s	F	Q^2	s-PRESS
attori.	Log1/C = $-0.330 (\pm 0.28) [\pi]^2 R_2 + 0.482 (\pm 0.42) \pi R_2 +$ 3.738 (±0.15)	23	0.490	0.259	3.156	-4.127	0.672
	$Log1/C = -0.254 (\pm 0.16) [\pi]^2 R_2 + 0.611 (\pm 0.25) \pi R_2 - 0.278$	23	0.872	0.149	.20.114	0.670	0.175
	$(\pm 0.090) B_4 R_2 + 4.173 (\pm 0.17)$						
	π -optimum = 1.20						
ble 3: Co	relation matrix of variables used in eq. 3. $\pi R_2 \qquad B_4 R_2$	abro T. anho					
ble 3: Co π R ₂		abro T unito T					

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